



(12) **EUROPEAN PATENT APPLICATION**
published in accordance with Art. 158(3) EPC

(43) Date of publication:
06.07.2005 Bulletin 2005/27

(51) Int Cl.7: **H01L 51/00**

(21) Application number: **03797648.7**

(86) International application number:
PCT/JP2003/011876

(22) Date of filing: **18.09.2003**

(87) International publication number:
WO 2004/027889 (01.04.2004 Gazette 2004/14)

(84) Designated Contracting States:
**AT BE BG CH CY CZ DE DK EE ES FI FR GB GR
HU IE IT LI LU MC NL PT RO SE SI SK TR**
Designated Extension States:
AL LT LV MK

(30) Priority: **19.09.2002 JP 2002273144**

(71) Applicant: **Daikin Industries, Ltd.**
Osaka-shi, Osaka 530-8323 (JP)

(72) Inventors:
• **Morita, Masamichi**
Sanda-shi, Hyogo 669-1348 (JP)
• **Otsuka, Hideyuki**
Fukuoka-shi, Fukuoka 812-0061 (JP)

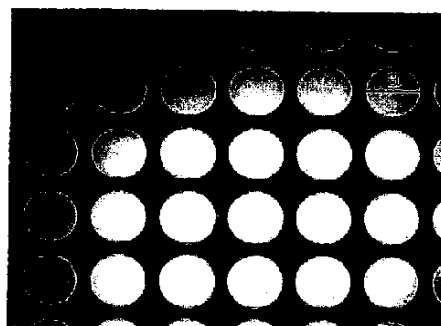
- **Takahara, Atsushi**
Fukuoka-shi, Fukuoka 814-0133 (JP)
- **Yamamoto, Ikuo,**
Yodogawa Works of DAIKIN IND. LTD.
Settsu-shi, Osaka 566-8585 (JP)
- **Itami, Yasuo,**
Yodogawa Works of DAIKIN IND. LTD.
Settsu-shi, Osaka 566-8585 (JP)
- **Aoyama, Hirokazu, Yodogawa Works**
Settsu-shi, Osaka 566-8585 (JP)

(74) Representative: **HOFFMANN - EITLE**
Patent- und Rechtsanwälte
Arabellastrasse 4
81925 München (DE)

(54) **MATERIAL WITH PATTERN SURFACE FOR USE AS TEMPLATE AND PROCESS FOR PRODUCING THE SAME**

(57) According to the present invention, there can be obtained an anisotropic material comprising an alternating-line pattern and a layer of a functional compound formed on the surface of the alternating-line pattern, wherein one type of lines is made of a fluorine-containing compound and the other type of lines is made of a non-fluorinated compound in the alternating-line surface. When using, as a template, a pattern surface produced by using a fluorine compound having a specific structure as a surface-treating agent, a structure of a functional compound of nanometer to micrometer order can be produced by a process of applying a functional compound solution. Properties of a functional compound can be improved by using, as a template, a pattern surface, at least one region of which is surface-treated with a fluorine compound having a specific structure.

Fig.1



a : 170nm
b : 30nm

Description

TECHNICAL FIELD

[0001] The present invention relates to a material with a pattern surface for used as a template, and a method for producing the same.

BACKGROUND ART

[0002] G. M. Whitesides et al. [Langmuir, 10, 1498 (1994)] showed that, when an aqueous $K_3Fe(CN)_6$ solution is applied on the surface of an alternating-line pattern comprising a non-fluorinated alkanethiol (hydrophobic region) and a carboxylic acid-terminated modified alkanethiol (hydrophilic region), $K_3Fe(CN)_6$ is crystallized on the hydrophilic region. However, there was such a problem that, when an organic solvent solution of a functional compound having low surface tension is applied on the surface of the line pattern of this combination, it is difficult to crystallize the functional compound in a line shape because the organic solvent solution spreads to wet both regions.

[0003] H. Sirringhaus et al. developed a technique wherein a partition measuring 50 nm in height and 5 μm in width made of hydrophobic polyimide is provided on a glass substrate and an aqueous conductive polymer solution is applied in the partition by an ink-jet method to form an electrically conductive polymer thin film having a width of 10 μm . However, this method had a problem that the process is complicated because the partition is provided.

[0004] JP-A-2002-261048 discloses a technique wherein a pattern surface having regions each having different affinity to a functional liquid is formed by using fluoroalkylsilane as a surface-treating agent and a functional liquid is applied to the region having high affinity by utilizing a difference in affinity. However, the fluoroalkylsilane are not described in detail in this document. Typical fluoroalkylsilane employed widely in general chemical industry is a compound having a straight-chain perfluoroalkyl group, which is represented by the formula:



[Langmuir, 8, 1195 (1992)] and fluoroalkylsilane having a long-chain perfluoroalkyl group ($n = 8, 10$) is used particularly popularly. However, the pattern surface surface-treated with the straight-chain perfluoroalkyltrichlorosilane had such a problem that a difference in the region coated selectively with the functional liquid is not clear.

[0005] JP-A-2000-307172, JP-A-7-206599 and JP-A-2001-94107 disclose techniques wherein performances of an organic semiconductor are improved by applying the organic semiconductor on a substrate surface-treated with a fluorine compound. However, there was such a problem that performances of the organic semiconductor are slightly improved.

DISCLOSURE OF THE INVENTION

(Technical Object to be Solved by the Invention)

[0006] An object of the present invention is to enable the production of an anisotropic material having an alternating-line pattern structure by a simple process of applying an organic solvent liquid. To achieve this object, it is necessary that a functional compound solution attains the wetting, which is faithful to an alternating-line pattern structure, on the surface of a substrate used as a template. That is, the functional compound solution spreads to wet one region of alternating lines on the substrate, and then a medium is vaporized to form a thin film of the functional compound in an alternating line shape.

[0007] Another object of the present invention is to produce a structure of a functional compound of nanometer to micrometer order by a process of applying a functional compound solution by using, as a template, a pattern surface composed of plural regions each having different surface free energy.

[0008] Still another object of the present invention is to apply a functional compound such as organic semiconductor to a pattern surface, at least one region of which is surface-treated with a fluorine compound, thereby improving properties of the functional compound.

(Means for Solving the Problems)

[0009] In a first aspect, the present invention provides an anisotropic material comprising an alternating-line pattern and a layer of at least one functional compound selected from the group consisting of a semiconductor compound, an electrically conductive compound, a photochromic compound and a thermochromic compound, formed on the surface

of the alternating-line pattern, wherein one type of lines comprises a first component comprising a fluorine-containing compound or silicone and the other type of lines comprises a second component comprising the other compound in the alternating-line pattern surface.

[0010] In a second aspect, the present invention provides a method for producing a functional material, comprising using, as a template, a pattern surface composed of plural regions each having different surface free energy, characterized in that:

- (1) at least one region of the pattern surface is treated with a fluorine compound, and
- (2) the method comprises applying a functional compound solution on the pattern surface and removing a solvent.

[0011] In a third aspect, the present invention provides a method for producing a functional material, which comprises applying a functional compound to a pattern surface having at least one region surface-treated with a fluorine compound.

BRIEF DESCRIPTION OF THE DRAWINGS

[0012]

- Fig. 1 is a view showing a circular photomask.
 Fig. 2 is a view showing a quadrangular photomask.
 Fig. 3 is a view showing a triangular photomask.
 Fig. 4 is a view showing a linear photomask.

MODE FOR CARRYING OUT THE INVENTION

[0013] When a solution prepared by dissolving a functional compound such as semiconductor compound, conductive compound, photochromic compound or thermochromic compound in an organic solvent is applied on the surface of an alternating-line pattern, a thin film of the functional compound is formed in an alternating line shape and therefore an anisotropic material (that is, a functional material) can be produced.

[0014] In the alternating-line pattern surface, one type of lines are made of a first component comprising a fluorine compound or silicone and the other one type of lines are made of a second component comprising the other compound.

[0015] Examples of the base material constituting the alternating-line pattern include silicon, synthetic resin, glass, metal, and ceramics.

[0016] The synthetic resin may be either a thermoplastic resin or a thermosetting resin and examples thereof include various thermoplastic elastomers, for example, polyolefins such as polyethylene, polypropylene, ethylene-propylene copolymer, and ethylene-vinyl acetate copolymer (EVA); cyclic polyolefin, modified polyolefin, polyvinyl chloride, polyvinylidene chloride, polystyrene, polyamide, polyimide, polyamideimide, polycarbonate, poly-(4-methylpentene-1), ionomer, acrylic resin, polymethyl methacrylate, acrylic-styrene copolymer (AS resin), butadiene-styrene copolymer, polyol copolymer (EVOH), polyesters such as polyethylene terephthalate (PET), polybutylene terephthalate (PBT) and polycyclohexane terephthalate (PCT); polyether, polyether ketone (PEK), polyether ether ketone (PEEK), polyether imide, polyacetal (POM), polyphenylene oxide, modified polyphenylene oxide, polyarylate, aromatic polyester (liquid crystal polymer), polytetrafluoroethylene, polyvinylidene fluoride, fluorine resin, and various thermoplastic elastomers such as styrene polymer, polyolefin, polyvinyl chloride, polyurethane, fluororubber and chlorinated polyethylene; and epoxy resin, phenol resin, urea resin, melamine resin, unsaturated polyester, silicone resin, polyurethane, and copolymer, polymer blend and polymer alloy composed mainly of these. One or more kinds of them can be used in combination (for example, as at least two-layered laminate).

[0017] Examples of the glass include silica glass (quartz glass), alkali-silicate glass, soda-lime glass, potash-lime glass, lead (alkali) glass, barium glass, and borosilicate glass.

[0018] Examples of the metal include gold, silver, copper, iron, nickel, aluminum, and platinum.

[0019] Examples of the ceramic include oxides (for example, aluminum oxide, zinc oxide, titanium oxide, silicon oxide, zirconia, and barium titanate), nitrides (for example, silicon nitride and boron nitride), sulfides (for example, cadmium sulfide), and carbides (for example, silicon carbide), and a mixture thereof may be used.

[0020] Even when any substrate is used, a functional group may be introduced into the surface of the substrate by plasma treatment or UV treatment.

[0021] The compound used to form the alternating-line pattern on the surface of the base material may be an organic silane compound, an organic thiol compound, an organic disulfide compound, or an organic phosphate ester compound. That is, lines made of a fluorine-containing compound or silicone and lines made of the other compound can be formed from an organic silane compound, an organic thiol compound, an organic disulfide compound, or an organic phosphate

ester compound.

[0022] A difference between surface free energy of lines made of a fluorine-containing compound or silicone and surface free energy of lines made of the other compound is preferably at least 5 mJ/m², for example, at least 10 mJ/m², particularly at least 20 mJ/m².

[0023] The line width of the alternating-line pattern may be from 0.5 μm to 100 μm, for example, from 1 μm to 20 μm. The line width may be constant or changed. The shape of lines may be either a straight line or a curved line.

[0024] The functional compound solution may be allowed to move spontaneously along lines by providing lines having lower surface free energy with tilt of surface free energy. Tilt of surface free energy can be provided, for example, by the method of M. K. Chaudhury et al. [Science, 256, 1539 (1992)].

[0025] Regarding unevenness of alternating lines, when the region having affinity with the functional compound is indented to form a groove, it is easy to wet in a line shape, as a matter of course. However, according to the present invention, it becomes possible to wet in a line shape by only a difference in surface free energy without providing unevenness intentionally. In case of preparing a pattern surface with only a monomolecular film, unevenness is not more than 10 nm, and preferably not more than 2 nm. The unevenness can be substantially neglected as compared with a line width of micron order. In the present invention, regardless of such a smooth surface, a solution of a functional compound spreads to wet one region of alternating lines, as if the groove existed.

[0026] In the present invention, the shape of droplets is preferably distorted longitudinally in a line direction when 2 μL of ethanol is gently dropped on a line pattern surface serving as a template used to prepare an anisotropic material. The degree of distortion is preferably at least 1.1, for example, at least 1.2, in terms of a ratio L/W of the length in a major axis (L) to the length in a minor axis (W) of droplets. When this condition is satisfied, the organic solvent remarkably tends to spread to wet one region of alternating lines on the alternating-line pattern surface.

[0027] Both lines of the alternating-line pattern surface used as a template may be made of a monomolecular film or not.

[0028] The alternating-line pattern surface of the present invention is prepared by the method such as photolithography, microcontact printing, self-assembling method, or electron beam irradiation method, and the method is not specifically limited. For example, Sugimura et al. [Langmuir, 16, 885 (2000)] reported that, when an organic silane monomolecular film formed on a silicon wafer as a substrate is irradiated with vacuum-ultraviolet light through a line-shaped photomask, the irradiated area is photodecomposed to form a silanol group, thus obtaining a line pattern surface in which organic silane regions and silanol group regions are alternately arranged. Furthermore, when another organic silane is chemically adsorbed on this surface, it selectively reacts with the silanol group region to give a composite monomolecular film (photolithography method). G. M. Whitesides et al. [Langmuir, 10, 1498 (1994)] reported that a line pattern surface can be prepared by applying alkanethiol as an ink to a silicone stamp and stamping on a gold substrate (microcontact printing method). M. Gleiche et al. ["Nanoscale channel lattices with controlled anisotropic wetting", NATURE, 403, 13 (2000)] reported that, when dipalmitoyl phosphatidylcholine (DPPC) as a hydrocarbon amphiphile spread over the water surface is transferred on a mica substrate under specific conditions, there can be obtained a line pattern surface in which DPPC and mica are alternately arranged in a self-assembling manner.

[0029] The alternating-line pattern is made of a first component and a second component.

[0030] The first component is a fluorine-containing compound or silicone. The second component is a compound other than the fluorine-containing compound and silicone.

[0031] Examples of the fluorine-containing compound include a fluorine-containing organic silane compound, a fluorine-containing organic thiol compound, a fluorine-containing organic disulfide compound, and a fluorine-containing organic phosphate ester compound.

[0032] Examples of the fluorine-containing group in the fluorine-containing compound include a fluoroalkyl group (particularly, a perfluoroalkyl group), and a perfluoropolyether group. The number of carbon atoms of the fluorine-containing group is normally from 1 to 20, for example, 1 to 10.

[0033] Examples of the silicone include dimethylpolysiloxane, methylphenylpolysiloxane, methylhydrogenpolysiloxane, and silicone resin, and these silicones may be modified with fluorine.

[0034] The compound (the second component) other than the fluorine compound and silicone may be an organic silane compound, an organic thiol compound, an organic disulfide compound, and/or an organic phosphate ester compound, which is not fluorinated or siliconated. The other compound may be a compound which has neither a fluorine atom nor a SiO bond.

[0035] On a line pattern surface, a layer of a functional compound is formed. The functional compound is a semiconductor compound, an electrically conductive compound, a photochromic compound, and/or a thermochromic compound.

[0036] The functional compound used in the present invention is at least one selected from the group consisting of the semiconductor compound, the conductive compound, the photochromic compound and the thermochromic compound.

[0037] The semiconductor compound is preferably an organic compound and examples thereof include a pentacene

derivative, a polythiophene derivative, a phthalocyanine derivative, a polyfluorene derivative, poly(p-phenylenevinylene), and a layered perovskite compound.

[0038] The conductive compound has conductivity of at least 10^2 S/cm at room temperature and is preferably an organic compound, and examples thereof include a polyacetylene derivative, a polythiophene derivative, polypyrrole, poly(p-phenylenevinylene), and polyaniline. The conductivity may be improved by doping these compounds.

[0039] The photochromic compound is preferably an organic compound and examples thereof include an azobenzene derivative, a spiropyran derivative, a fulgide derivative, and a diarylethene derivative.

[0040] The thermochromic compound is a generic term of a compound wherein a color of a substance varies reversibly with a change in temperature, and examples thereof include salicylideneanilines, a polythiophene derivative, a tetrahalogeno complex, an ethylenediamine derivative complex, a dinitrodiammine-copper complex, a 1,4-ziazacyclooctane (daco) complex, a hexamethylenetetramine (hmta) complex, and a salicylaldehyde (salen) complex.

[0041] The thickness of a layer of the functional compound may be from 0.1 nm to 100 μ m, for example, from 1 nm to 1 μ m.

[0042] The layer of the functional compound can be formed by applying a solution prepared by dissolving the functional compound in a solvent on the surface of a line pattern and removing the solvent. Examples of the solvent include an organic solvent and water. In case that the functional compound is slightly soluble in water, it must be dissolved in the organic solvent.

[0043] In the present invention, the solvent used to dissolve the functional compound is preferably a solvent having a surface tension of not more than 30 mN/m, for example, not more than 20 mN/m. When the surface tension is not more than 30 mN/m, the solution easily spreads to wet along lines.

[0044] Examples of the organic solvent include alcohols, esters, ketones, ethers, and hydrocarbons (for example, aliphatic hydrocarbons and aromatic hydrocarbons) and the organic solvent may be fluorinated or not. Specific examples of the organic solvent include methanol, ethanol, isopropanol, perfluorodecalin, hydrofluoroether, HCFC225, chloroform, 1,1,2,2-tetrachloroethane, tetrachloroethylene, chlorobenzene, ethyl acetate, butyl acetate, acetone, hexane, isopentane, toluene, xylene, and tetrahydrofuran.

[0045] The concentration of the functional compound in the solution may be from 0.1 to 20% by weight, for example, from 1 to 10% by weight.

[0046] The solvent can be removed by vaporization. The solvent can be removed by heating an anisotropic material, for example, to a temperature of 60 to 200°C. The solvent may be removed under reduced pressure (for example, 0.01 to 100 Pa).

[0047] In the present invention, examples of the method of applying the functional compound solution include a spin coating method, a dip coating method, a casting method, a roll coating method, a printing method, a transfer method, an ink-jet printing method [P. Calvert, Chem. Mater., 13, 3299 (2001)], a bar coating method, and a capillary method.

[0048] The anisotropic material (that is, the functional material) of the present invention can be used as an electronic device in transistors, memories, light emitting diodes (EL), lasers and solar batteries, and is also used as an optical device in optical memories, image memories, light modulation elements, optical shutters, secondary harmonic (SHG) devices and polarizers.

[0049] In the second aspect of the present invention, it has been found that a functional material is produced by a method wherein a pattern surface prepared by using the fluorine compound as a surface-treating agent is used as a template and a functional compound solution is applied, and then a solvent is removed. Thus, a functional material as a structure of a functional compound of nanometer to micrometer order, for example, size of about 10 nm to 100 μ m (for example, width (particularly, maximum width or minimum width) or diameter) is obtained.

[0050] In this second aspect, descriptions are the same as those in the first aspect, except that the pattern shape is not limited to line, the functional compound is not limited to a semiconductor compound, an electrically conductive compound, a photochromic compound or a thermochromic compound, and a fluorine compound is used to prepare a pattern surface.

[0051] The shape of the pattern surface may be selected according to the purposes of the finally produced device. Examples of the shape include circle, quadrangle, triangle, straight line, and curved line. Patterns may be contacted or separated with each other. The pattern surface is characterized by comprising at least two kinds of regions each having different surface free energy, at least one region being surface-treated with a fluorine compound.

[0052] Examples of the functional compound include a semiconductor compound, a conductive compound, a photochromic compound, a thermochromic compound, a magnetic compound, and a bioactive compound.

[0053] The fluorine compound used to prepare a pattern surface is preferably a fluorine compound having the following structures:

- (a) a fluorine compound which has a branched fluoroalkyl group having 5 or less carbon atoms,
- (b) a fluorine compound having a perfluoropolyether group,
- (c) a fluorine compound having a polymer structure obtained by polymerizing a monomer which has a fluoroalkyl

group having 5 or less carbon atoms,

(d) a fluorine compound having a linking group which is any one of an urethane group, an ester group, an ether group and an amide group, existing between a fluoroalkyl group having 5 or less carbon atoms and a functional group,

(e) an incompletely-condensed silsesquioxane which has a fluoroalkyl group having 5 or less carbon atoms, and/or
(f) a completely-condensed silsesquioxane which has a silane group and a fluoroalkyl group having 5 or less carbon atoms.

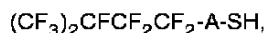
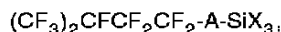
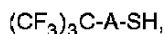
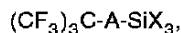
[0054] The fluoroalkyl group in the fluorine compound may be a perfluoroalkyl group (having 1 to 5 carbon atoms).

[0055] Examples of the fluorine compound include a fluorine-containing organic silane compound, a fluorine-containing organic thiol compound, a fluorine-containing organic disulfide compound, and a fluorine-containing organic phosphate ester compound.

[0056] The fluorine compound has a functional group, for example, a silane group ($-SiX_3$ (wherein X is a hydrogen atom, a halogen atom or an oxyalkyl group (having 1 to 4 carbon atoms))), a thiol group ($-SH$), a disulfide group ($-S-S-$), or a phosphoric acid group ($P(=O)(OH)_{3-n}(O-)_n$) (provided that n is 1 to 3).

[0057] In the fluorine compound (a) which has a branched fluoroalkyl group having 5 or less carbon atoms, examples of the branched fluoroalkyl group include $(CF_3)_2CF-$, $(CF_3)_3C-$, $(CF_3)_2CFCF_2CF_2-$, and $(CF_3)_3CCF_2-$. The fluorine compound (a) is a compound having a branched fluoroalkyl group and a functional group (for example, a silane group).

[0058] Examples of the fluorine compound (a) include the followings:

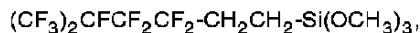
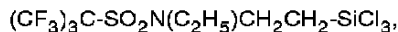
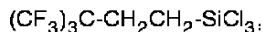


and

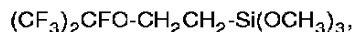


wherein A is an alkylene group having 1 to 4 carbon atoms, a $-SO_2N(R^{21})R^{22}-$ group (provided that R^{21} is an alkyl group having 1 to 4 carbon atoms, and R^{22} is an alkylene group having 1 to 4 carbon atoms) or a $-CH_2CH(OH)CH_2-$ group, and X is a halogen atom (for example, a chlorine atom), or OC_nH_{2n+1} ($n=1$ to 4).

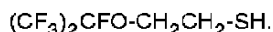
[0059] Specific examples thereof include the followings:



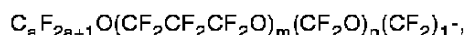
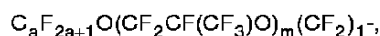
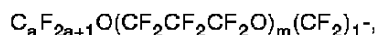
EP 1 551 066 A1



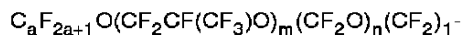
and



[0060] In the fluorine compound (b) having a perfluoropolyether group, examples of the perfluoropolyether group include the following:

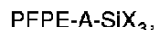


and

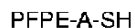


wherein a molecular weight of a perfluoropolyether group is at least 300, preferably at least 500, and not more than 100000, for example, not more than 5000, m and n represent 1 to 100, and preferably 20 to 40, a represents 1 to 10, and 1 represents 1 or 2. The fluorine compound (b) is a compound having a perfluoropolyether group and a functional group (for example, a silane group).

[0061] Examples of the fluorine compound (b) include the followings:

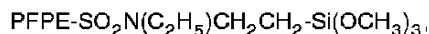
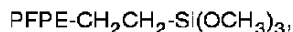


and

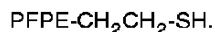


wherein PFPE is a perfluoropolyether group, A is an alkylene group having 1 to 4 carbon atoms, a $-\text{SO}_2\text{N}(\text{R}^{21})\text{R}^{22}$ -group (provided that R^{21} is an alkyl group having 1 to 4 carbon atoms, and R^{22} is an alkylene group having 1 to 4 carbon atoms) or a $-\text{CH}_2\text{CH}(\text{OH})\text{CH}_2$ - group, and X is a halogen atom (for example, chlorine atom), or $\text{OC}_n\text{H}_{2n+1}$ ($n=1$ to 4).

[0062] Specific examples thereof include the followings:



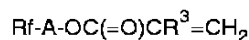
and



[0063] In the fluorine compound (c) having a polymer structure obtained by polymerizing a monomer which has a fluoroalkyl group having 5 or less carbon atoms, examples of the fluoroalkyl group having 5 or less carbon atoms include CF_3 -, CF_3CF_2 -, $\text{CF}_3\text{CF}_2\text{CF}_2$ -, $(\text{CF}_3)_2\text{CF}$ -, $\text{CF}_3\text{CF}_2\text{CF}_2\text{CF}_2$ -, $(\text{CF}_3)_2\text{CFCF}_2$ -, $(\text{CF}_3)_3\text{C}$ -, $\text{CF}_3\text{CF}_2\text{CF}_2\text{CF}_2\text{CF}_2$ -, $(\text{CF}_3)_2\text{CFCF}_2\text{CF}_2$ -, and $(\text{CF}_3)_3\text{CCF}_2$ -. The monomer having a fluoroalkyl group is a compound having a fluoroalkyl group and a carbon-carbon double bond. The monomer having a fluoroalkyl group may be a fluoroalkyl group-containing vinyl monomer.

[0064] The fluoroalkyl group-containing vinyl monomer may be a (meth)acrylate having a fluoroalkyl group, or an acrylate in which hydrogen at the α -position having a fluoroalkyl group is substituted with a halogen atom, CF_3 , CF_2H or CFH_2 .

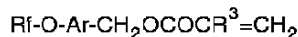
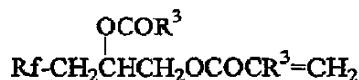
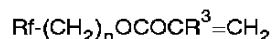
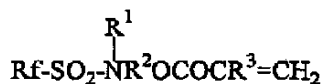
[0065] The fluoroalkyl group-containing acrylate may be represented by the following general formula:



wherein Rf is a straight-chain or branched fluoroalkyl group having 1 to 5 carbon atoms, R^3 is a hydrogen atom, an F atom, a Cl atom, a CF_3 group, a CF_2H group, a CFH_2 group or a methyl group, and A is a divalent organic group.

[0066] Examples of A include an alkylene group having 1 to 4 carbon atoms, a $-\text{SO}_2\text{N}(\text{R}^{21})\text{R}^{22}$ - group (provided that R^{21} is an alkyl group having 1 to 4 carbon atoms, and R^{22} is an alkylene group having 1 to 4 carbon atoms) and a $-\text{CH}_2\text{CH}(\text{OH})\text{CH}_2$ - group.

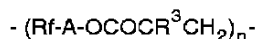
[0067] Examples of the fluoroalkyl group-containing acrylate include the followings:



wherein Rf is a fluoroalkyl group having 1 to 5 carbon atoms, R^1 is a hydrogen or an alkyl group having 1 to 10 carbon atoms, R^2 is an alkylene group having 1 to 10 carbon atoms,

R^3 is a hydrogen atom, an F atom, a Cl atom, a CF_3 group, a CF_2H group, a CFH_2 group or a methyl group, Ar is an arylene group which may have a substituent, and n is an integer of 1 to 10.

[0068] Examples of the polymer structure obtained by polymerizing a monomer which has a fluoroalkyl group having 5 or less carbon atoms include the followings:



EP 1 551 066 A1

wherein Rf, A and R³ are as defined above, and n is from 1 to 200, and particularly from 2 to 100.

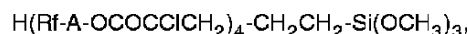
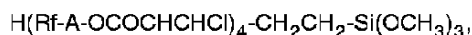
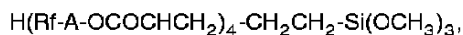
[0069] Examples of the fluorine compound (c) include the followings:

Polymer-D-SiX₃, and
Polymer-D-SH

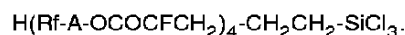
wherein Polymer represents a polymer structure group obtained by polymerizing a monomer which has a fluoroalkyl group having 5 or less carbon atoms, D represents a divalent group obtained by linking S and an urethane group interposed between two alkylene groups having 1 to 4 carbon atoms, for example, -S(CH₂)₂OCONH(CH₂)₃-, and X

represents a halogen atom (for example, a chlorine atom), or OC_nH_{2n+1} (n=1 to 4).

[0070] Specific examples thereof include the followings:

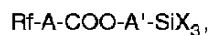
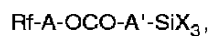


and



[0071] In the fluorine compound (d) having a linking group which is any one of an urethane group, an ester group, an ether group and an amide group, existing between a fluoroalkyl group having 5 or less carbon atoms and a functional group, examples of the fluoroalkyl group having 5 or less carbon atoms are the same as those in the fluorine compound (c).

[0072] Examples of the fluorine compound (d) include the followings:



EP 1 551 066 A1



5



10



and

15

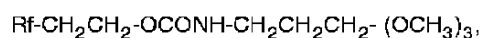


wherein Rf represents a fluoroalkyl group having 5 or less carbon atoms, A and A' represent an alkylene group having 1 to 4 carbon atoms, a $-\text{SO}_2\text{N}(\text{R}^{21})\text{R}^{22}-$ group (provided that R^{21} is an alkyl group having 1 to 4 carbon atoms, and R^{22} is an alkylene group having 1 to 4 carbon atoms) or a $-\text{CH}_2\text{CH}(\text{OH})\text{CH}_2-$ group, and X is a halogen atom (for example, chlorine atom), or $\text{CC}_n\text{H}_{2n+1}$ ($n=1$ to 4).

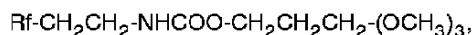
20

[0073] Examples thereof include the followings:

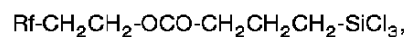
25



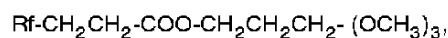
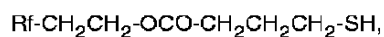
30



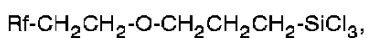
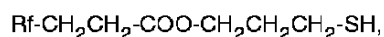
35



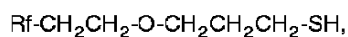
40



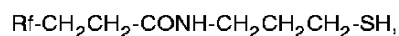
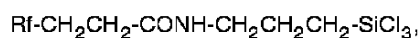
45

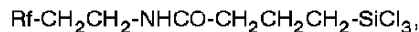


50

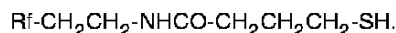


55

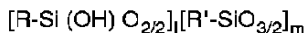




and

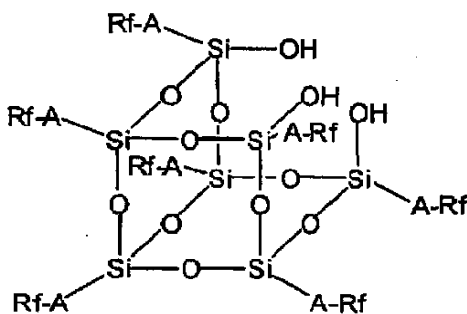
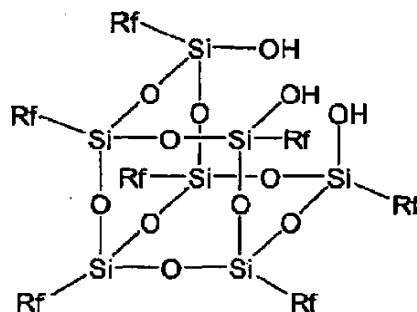


[0074] In the incompletely-condensed silsesquioxane (e) which has a fluoroalkyl group having 5 or less carbon atoms, examples of the fluoroalkyl group having 5 or less carbon atoms are the same as those in the fluorine compound (c). The incompletely-condensed silsesquioxane (e) may be represented by the general formula:



wherein R and R' represent Rf, Rf-A, an alkyl group (having 1 to 22 carbon atoms), or a derivative of an alkyl group (having 1 to 22 carbon atoms) (provided that at least one of R and R' is Rf or Rf-A) (wherein Rf represents a fluoroalkyl group having 5 or less carbon atoms, and A represents an alkylene group having 1 to 4 carbon atoms, a $-\text{SO}_2\text{N(R}^{21})\text{R}^{22}-$ group (provided that R²¹ is an alkyl group having 1 to 4 carbon atoms, and R²² is an alkylene group having 1 to 4 carbon atoms) or a $-\text{CH}_2\text{CH(OH)CH}_2-$ group), and l and m represent such a number that a molecular weight of the incompletely-condensed silsesquioxane is within a range from 500 to 100000. The silsesquioxane is also referred to as POSS (Polyhedral Oligomeric Silsesquioxane). A silanol group of the incompletely-condensed silsesquioxane has high reactivity and is therefore linked strongly with an active OH group of the substrate. The number of the silanol group in the molecule is from 1 to 3. l is normally from 1 to 3. m is normally at least 1, for example, at least 2, and particularly at least 3.

[0075] Specific examples of the incompletely-condensed silsesquioxane (e) include:



Structural formula (1)

and POSS Silanols 54,456-6 manufactured by Aldrich Co.

[0076] In the completely-condensed silsesquioxane (f) which has a silane group and a fluoroalkyl group having 5 or less carbon atoms, the silane group is a $-SX_3$ group (X is a halogen atom (for example, a chlorine atom), or OC_nH_{2n+1} (n=1 to 4)). Examples of the fluoroalkyl group having 5 or less carbon atoms are the same as those in the fluorine compound (c).

[0077] The completely-condensed silsesquioxane (f) may be represented by the formula:



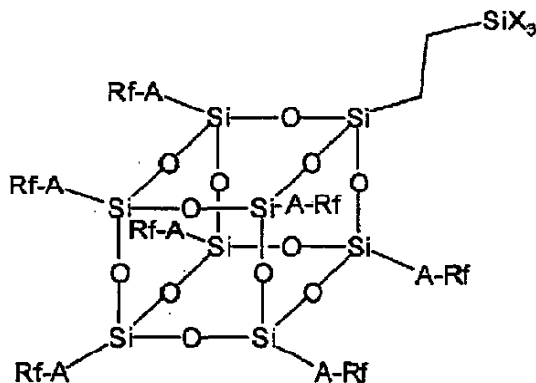
wherein R represents Rf, Rf-A, an alkyl group (having 1 to 22 carbon atoms), or a derivative of an alkyl group (having 1 to 22 carbon atoms) (provided that at least one of R is Rf or Rf-A) (wherein Rf represents a fluoroalkyl group having 5 or less carbon atoms, A represents an alkylene group having 1 to 4 carbon atoms, a $-SO_2N(R^{21})R^{22}-$ group (provided that R^{21} is an alkyl group having 1 to 4 carbon atoms, and R^{22} is an alkylene group having 1 to 4 carbon atoms) or a $-CH_2CH(OH)CH_2-$ group), R' represents an organic group containing SiX_3 , for example, $-(CH_2)_nSiX_3$ (n is from 1 to 10), and l and m represent such a number that a molecular weight of the completely-condensed silsesquioxane is within a range from 500 to 100000.

[0078] l and m are normally at least 1.

[0079] The completely-condensed silsesquioxane (f) can be obtained by condensing an incompletely-condensed silsesquioxane (e) with bis-silane. Examples of the bis-silane include the followings:

bis-(triethoxysilyl)ethane,
bis-(triethoxysilylpropyl)amine, and
bis-(trimethoxysilylpropyl)tetrasulfide.

[0080] Specific examples of the completely-condensed silsesquioxane (f) include the following:



Structural formula (2)

wherein X is a halogen atom (for example, a chlorine atom), or OC_nH_{2n+1} (n=1 to 4).

[0081] In the third aspect of the present invention, a functional material is obtained by using, as a template, a pattern surface, at least one region of which is surface-treated with a fluorine compound, and applying a functional compound such as organic semiconductor on the template. Thus, properties of the functional compound can be improved.

[0082] In the third aspect, explanations are the same as those recited in the second aspect, except that the method of applying the functional compound to the pattern surface is not limited to an application treatment.

[0083] The method of applying the functional compound to the pattern surface is not specifically limited as far as it is a method of contacting the functional compound with the pattern surface, and a method capable of improving properties of the functional compound as highly as possible may be appropriately selected regardless of solid phase, liquid

phase or vapor phase. Examples of the application method include a friction transfer method, a spin coating method, a casting method, an ink-jet method, a dipping method, a vacuum deposition method, and a sputtering method.

[0084] Even if the functional compound is applied on the whole surface of the pattern surface, a difference in structure of a thin film made of the functional compound arises in the region surface-treated with the fluorine compound and the other region, and thus properties of the functional compound can be maximized.

PREFERRED EMBODIMENTS

[0085] The present invention will now be described by way of examples, but is not limited to the following examples.

1. Test Example in which a fluorine compound is used for one alternating-line pattern surface

• Preparation of alternating-line pattern surface

[0086] After a silicon wafer is washed with acetone, the surface is terminated with a hydroxyl group by heating in hydrogen peroxide water/concentrated sulfuric acid = 3/7 (volume ratio) at 110°C for one hour. An organic silane as a first component is chemically adsorbed on the silicon wafer. When the silicon wafer is irradiated with vacuum-ultraviolet light (172 nm) through a line-shaped photomask (line width: 0.1 μm , 1 μm , 5 μm , and 10 μm), the irradiated organic area is photodecomposed to form a silanol group, thus obtaining a line pattern surface in which organic silane regions and silanol group regions are alternately arrayed. When another organic silane as a second component is chemically adsorbed on the surface, it selectively reacts with the silanol group region to give a composite monomolecular film. According to the above method, pattern surfaces having various combinations of a fluoroalkyl group (C_6F_{13}), an alkyl group ($\text{C}_{10}\text{H}_{21}$ and $\text{C}_{18}\text{H}_{37}$), a silanol group (SiOH) and a sulfonic acid group (SO_3H) were prepared.

[0087] Hereinafter, the respective surfaces are abbreviated as follows: $\text{C}_6\text{F}_{13} \rightarrow \text{Rf}$, $\text{C}_{10}\text{H}_{21} \rightarrow \text{Rh}(\text{C}_{10})$ and $\text{C}_{18}\text{H}_{37} \rightarrow \text{Rh}(\text{C}_{18})$, and SiOH and SO_3H are used without being abbreviated.

[0088] Reagents used to prepare the respective surfaces are as follows: Rf is perfluorohexylethyltrimethoxysilane (manufactured by Fluorochem Ltd.), $\text{Rh}(\text{C}_{18})$ is n-octadecyltrimethoxysilane (manufactured by CHISSO CORPORATION), and $\text{Rh}(\text{C}_{10})$ is n-decyltriethoxysilane (manufactured by CHISSO CORPORATION). In case of SO_3H , γ -mercaptopropyltrimethoxysilane (manufactured by CHISSO CORPORATION) was chemically adsorbed, irradiated with UV light (254 nm) for 10 hours, thereby converting a terminal SH group into a sulfonic acid group due to photooxidation.

[0089] Formation of the objective pattern was confirmed by a lateral force microscope.

• Method for measurement of surface free energy of sole monomolecular film

[0090] Surface free energy of a sole monomolecular film, which is not a patterning surface, was determined by substitution of a contact angle of water and methylene iodide into the formula of D. K. Owens [J. Applied Polym. Sci., 13, 1741 (1969)] (Table 1).

Table 1:

Contact angle and surface free energy on various sole monomolecular films						
Monomolecular film	Water	Methylene iodide	Surface free energy (water-methylene iodide)			
			γ_d	γ_P	γ	Difference from Rf
SiOH	0	0	38.8	37.5	76.3	62
SO_3H	40	42	29.5	29.9	59.4	45
$\text{Rh}(\text{C}_{18})$	103	61	28.3	0.2	28.5	14
$\text{Rh}(\text{C}_{10})$	97	68	22.4	2.0	24.4	10
Rf	106	88	12.3	1.9	14.2	-

• Pattern surface prepared

[0091] Pattern surfaces shown in Table 2 were prepared. In the column of "combination of patterns", "first component/second component" was described. "Difference in surface free energy" is a difference in surface free energy of a sole monomolecular film determined in Table 1. "Line width" was measured by a lateral force microscope. "Unevenness of alternating lines" denotes a difference between a molecular length of the first component and a molecular length of

EP 1 551 066 A1

the second component, and was measured by an atomic force microscope. Regarding "L/S of ethanol droplets ($2\text{ }\mu\text{L}$)", the degree of distortion, when $2\text{ }\mu\text{L}$ of ethanol is gently dropped from above the pattern surface, is expressed by a ratio L/W of the length in a major axis (L) to the length in a minor axis (W) of droplets. "Wetting of n-hexadecane in alternating line shape" was observed by an optical microscope after spin coating with a 1 % ethanol solution (2000 rpm). The case in which regions where n-hexadecane spreads to wet in a line shape and regions where no line exists are alternately formed was rated "Good", while the case where n-hexadecane spreads to wet in an indeterminate form was rated "Poor".

10

15

20

25

30

35

40

45

50

55

Table 2: Pattern surface

Examples	Combination of patterns	Difference in surface free energy (mJ/m ²)	Line width (μm)	Unevenness of alternating lines (nm)	L/W of ethanol droplets (2 μL)	Wetting of n-hexadecane in alternating-line shape
Preparation Example 1	Rf/SiOH	62	1.0	0.8	1.22	Good
Preparation Example 2	Rf/SO ₃ H	45	5.0	0.0	1.20	Good
Preparation Example 3	Rf/SO ₃ H	45	10.0	0.0	1.58	Good
Preparation Example 4	RE/Rh(C18)	14	5.0	1.5	1.60	Good
Preparation Example 5	RE/Rh(C10)	10	10.0	0.5	1.22	Good
Comparative Preparation Example 1	Rf alone	—	—	0.0	1.00	Poor
Comparative Preparation Example 2	Rh(C10)/SO ₃ H	35	10.0	0.3	impossible to measure*	Poor
Comparative Preparation Example 3	RE/Rh(C18)	14	0.1	1.5	↑	Poor
Comparative Preparation Example 4	Rh(C10)/Rh(C18)	4	5.0	1.3	↑	Poor

*: Ethanol spreads to wet the entire surface of substrate

Examples 1 to 5 and Comparative Examples 1 to 4

[0092] Using each pattern surface shown in Table 2 as a template, a 1 wt% functional compound solution shown in Table 3 was applied thereon. The spin coating was performed at 2000 rpm, and the dip coating was performed by fixing a substrate in a direction in which lines of a pattern are vertical to the solution surface, and the dipping and pull-up

operations were performed at a rate of 1 mm/second.

Table 3

Example	Pattern surface used	Combination of patterns	Difference in surface free energy (mJ/m ²)	Line width (μm)	Unevenness of alternating lines (nm)
Example 1	Preparation Example 1	Rf/SiOH	62	1.0	0.8
Example 2	Preparation Example 2	Rf/SO ₃ H	45	5.0	0.0
Example 3	Preparation Example 3	Rf/SO ₃ H	45	10.0	0.0
Example 4	Preparation Example 4	Rf/Rh (Cl8)	14	5.0	1.5
Example 5	Preparation Example 5	Rf/Rh (Cl10)	10	10.0	0.5
Comparative Example 1	Comparative Preparation Example 1	Rf alone	—	10.0	0.0
Comparative Example 2	Comparative Preparation Example 2	Rh(Cl10)/SO ₃ H	35	10.0	0.3
Comparative Example 3	Comparative Preparation Example 3	Rf/Rh (Cl8)	14	0.1	1.5
Comparative Example 4	Comparative Preparation Example 4	Rh(Cl10)/Rh(Cl8)	4	5.0	1.3

Table 3 (continued)

Example	Functional material applied			
	Spin coating of poly(dioctylfluorene)/xylene solution	Dip coating of poly(3-hexylthiophene)/chloroform solution	Spin coating of 6AZ10-PVA/chloroform solution	Dip coating of N-(5-chloro-2-hydroxybenzylidene)-aniline/isopentane solution
Example 1	Yes	-	-	-
Example 2		Yes	-	-
Example 3	-	-	Yes	
Example 4		-	-	Yes
Example 5	Yes	-	-	-
Comparative Example 1	Yes	-	-	-
Comparative Example 2	-	Yes	-	-
Comparative Example 3	-	-	Yes	-
Comparative Example 4	-	-	-	Yes

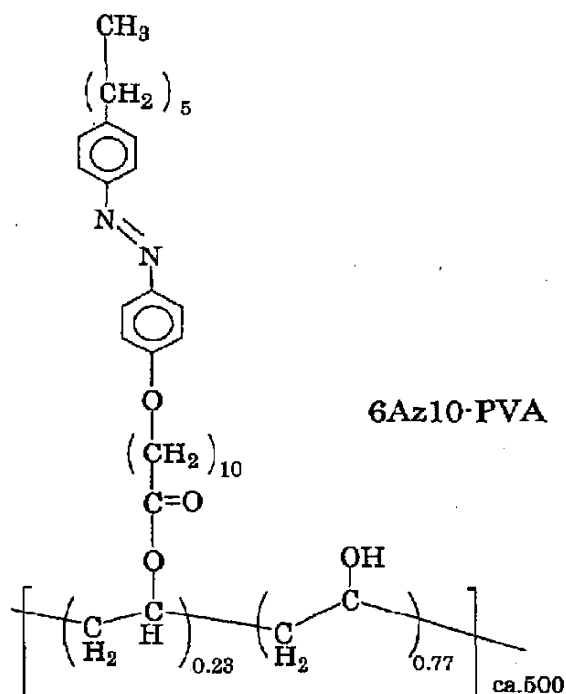
[0093] Functional materials shown in Table 3 are as follows.

Poly(dioctylfluorene): ADS129BE, manufactured by American Dye Source, Inc.

Poly(3-hexylthiophene): manufactured by Rieke Metals Co. (Aldrich Product No. 44570-3), head-to-tail regiospe-

cific: at least 98.5%

6Az10-PVA: which has the following structure in which azobenzene is introduced into a polyvinyl alcohol side chain, and synthesized by the method of T. Seki et al. [J. Phys. Chem. B, 103, 10338 (1999)]



N-(5-chloro-2-hydroxybenzylidene)-aniline: which is synthesized by the method of Konawa et al. [J. Am. Chem. Soc., 120, 7107 (1998)]

[0094] In Examples 1 to 5 according to the present invention, it was confirmed by an optical microscope and an atomic force microscope that a thin film (120 nm thick) of a functional compound is formed in an alternating line shape on the pattern surface used as the template. In Comparative Examples 1 to 4, a uniform thin film (140 nm thick) of a functional compound was formed on the entire surface of the substrate without constituting a line.

2. Test Example in which silicone is used for one alternating-line pattern surface

Preparation Example 6 and Example 6

[0095] A stamp having a line width of 5 μm , the surface of which is coated by the method of T. Deng et al. [Langmuir, 18, 5720(2002)], was prepared. A 0.5 wt% dimethylpolysiloxane (100 cSt)/toluene solution was applied to the stamp and then applied on a glass substrate. Formation of an alternating-line pattern was confirmed by an atomic force microscope. As a result, it has been found that a dimethylpolysiloxane thin film having a width of 5 μm is formed and the thickness of the film is 50 nm. On the pattern surface, a 1% aqueous solution of sodium poly[2-(3-thienyl)ethoxy-4-butylsulfonate] (ADS2000P, manufactured by American Dye Source, Inc.) was spin-coated (2000 rpm). It was confirmed by a fluorescence microscope and an atomic force microscope that a thin film (170 nm thick) of ADS2000P is formed on the pattern surface in an alternating line shape.

3. Test Example in which a fluorine compound having a specific structure is used for one pattern surface with various shapes

Preparation Examples 7 to 17 and Comparative Preparation

Example 5

[0096] On the same silicon wafer as in Preparation Example 1, a fluorine organic silane shown in Table 4 was chemically adsorbed in a liquid phase by the following method. The organic silane was diluted with a soluble dehydrated organic solvent, and then a silicon wafer was dipped in the resulting 1 wt% solution for one hour. After dipping, the silicon wafer was washed with the same organic solvent and dried in an atmospheric air to prepare a sole monomolecular film of the fluorine organic silane alone. A contact angle and surface free energy of the sole monomolecular film, which is not a pattern surface, were also shown in Table 4.

Table 4: Contact angle and surface free energy on various sole monomolecular films

Abbreviations	Organic silane	Water	Methylene iodide	Surface free energy (water-methylene iodide)		
				Yd	YP	Y
Br-Rf1	$(\text{CF}_3)_3\text{C}-\text{CH}_2\text{CH}_2-\text{Si}(\text{OCH}_3)_3$	118	95	10.2	0.4	10.6
Br-Rf2	$(\text{CF}_3)_2\text{CFCF}_2\text{CF}_2-\text{CH}_2\text{CH}_2-\text{Si}(\text{OCH}_3)_3$	116	95	10.0	0.6	10.7
Br-Rf3	$(\text{CF}_3)_2\text{CFO}-\text{CH}_2\text{CH}_2\text{CH}_2-\text{SiCl}_3$	115	94	10.4	0.7	11.1
PFPE	$\text{C}_3\text{F}_7(\text{OCF}_2\text{CF}_2\text{CF}_2)_n-\text{OCF}_2\text{CF}_2-\text{CH}_2\text{CH}_2-\text{Si}(\text{OCH}_3)_3$, $n=20$	123	101	8.1	0.2	8.3
Rf-Ac	$\text{H}[(\text{CF}_3(\text{CF}_2)_3-\text{CH}_2\text{CH}_2-\text{OCOCHCH}_2)_n-\text{SCH}_2\text{CH}_2\text{OCONH}(\text{CH}_2)_3-\text{Si}(\text{OCH}_3)_3]$	119	93	11.2	0.2	11.4
Rf-Ur	$(\text{CF}_3(\text{CF}_2)_3-\text{CH}_2\text{CH}_2-\text{OCONH}-(\text{CH}_2)_3-\text{Si}(\text{OCH}_3)_3)$	117	95	10.1	0.5	10.6
Rf-Es	$(\text{CF}_3(\text{CF}_2)_3-\text{CH}_2\text{CH}_2-\text{OCO}-(\text{CH}_2)_3-\text{Si}(\text{OCH}_3)_3)$	118	96	9.8	0.4	10.2
Rf-Et	$(\text{CF}_3(\text{CF}_2)_3-\text{CH}_2\text{CH}_2-\text{O}-(\text{CH}_2)_3-\text{Si}(\text{OCH}_3)_3)$	119	98	9.0	0.4	9.5
Rf-Am	$(\text{CF}_3(\text{CF}_2)_3-\text{CH}_2\text{CH}_2-\text{CONH}-(\text{CH}_2)_3-\text{SiX}_3)$	115	94	10.4	0.7	11.1
Rf-POSS1	A is CH_2CH_2 in the above structural formula (1)	123	102	6.6	0.4	7.0
Rf-POSS2	A is CH_2CH_2 and X is OC_2H_5 in the above structural formula (2)	122	100	6.2	0.6	6.8
St-Rf	$\text{CF}_3(\text{CF}_2)_3-\text{CH}_2\text{CH}_2-\text{Si}(\text{OCH}_3)_3$	93	72	19.5	3.8	23.3

[0097] Then, the resulting sole monomolecular film was irradiated with vacuum-ultraviolet light (wavelength: 172 nm) through a photomask having each shape shown in Figs. 1 to 4 (circle, quadrangle, triangle and line) to form a silanol group in the irradiated area. In Figs. 1 to 4, dark areas represent the masked regions. The resulting pattern surface is shown in Table 5. Formation of the objective pattern was confirmed by a field emission scanning electron microscope

(FE-SEM).

Table 5

	Organic silane	Combination of pattern	Shape of pattern
Preparation Example 7	Br-Rf1	Br-Rf1/SiOH	circle
Preparation Example 8	Br-Rf2	Br-Rf2/SiOH	line
Preparation Example 9	Br-Rf3	Br-Rf3/SiOH	triangle
Preparation Example 10	PFPE	PFPE/SiOH	line
Preparation Example 11	Rf-Ac	Rf-Ac/SiOH	circle
Preparation Example 12	Rf-Ur	Rf-Ur/SiOH	line
Preparation Example 13	Rf-Es	Rf-Es/SiOH	triangle
Preparation Example 14	Rf-Et	Rf-Et/SiOH	line
Preparation Example 15	Rf-Am	Rf-Am/SiOH	circle
Preparation Example 16	Rf-POSS1	Rf-POSS1/SiOH	quadrangle
Preparation Example 17	Rf-POSS2	Rf-POSS2/SiOH	line
Comparative Preparation Example 5	St-Rf	St-Rf/SiOH	line

Examples 7 to 17 and Comparative Example 5

[0098] Using each pattern surface shown in Table 5 as a template, a 1 wt% functional compound solution shown in Table 6 was applied thereon. Spin coating was performed at 2000 rpm, and dip coating was performed by pulling-up the pattern surface at a rate of 1 mm/second.

Table 6

Example	Pattern surface used	Combination of patterns	Functional material applied			
			Spin coating of poly(dioctylfluorene)/xylene solution	Dip coating of poly(3-hexylthiophene)/chloroform solution	Spin coating of 6Az10-PVA/chloroform solution	Dip coating of N-(5-chloro-2-hydroxybenzylidene)-aniline/isopentane solution
Example 7	Preparation Example 7	Br-Rf1/SiOH	Yes	-	-	-
Example 8	Preparation Example 8	Br-Rf2/SiOH	-	Yes	-	-
Example 9	Preparation Example 9	Br-Rf3/SiOH	-	Yes	-	-
Example 10	Preparation Example 10	PFPE/SiOH	-	-	-	Yes
Example 11	Preparation Example 11	Rf-Ac/SiOH	-	-	-	Yes
Example 12	Preparation Example 12	Rf-Ur/SiOH	-	Yes	-	-
Example 13	Preparation Example 13	Rf-Es/SiOH	-	-	Yes	-
Example 14	Preparation Example 14	Rf-Et/SiOH	Yes	-	-	-
Example 15	Preparation Example 15	Rf-Am/SiOH	Yes	-	-	-
Example 16	Preparation Example 16	Rf-POSS1/SiOH	-	-	Yes	-
Example 17	Preparation Example 17	Rf-POSS2/SiOH	-	Yes	-	-
Comparative Example 5	Comparative Preparation Example 5	st-Rf/SiOH	Yes	-	-	-

[0099] In Examples 7 to 17 of the present invention, it was confirmed by an optical microscope and an atomic force microscope that a thin film of a functional compound is formed in a shape, which is the same as that of the pattern surface, on the pattern surface used as the template (150 nm thick). The thin film was formed only in the SiOH region. In Comparative Example 5, a uniform thin film (140 nm thick) of the functional compound was formed on the entire surface of the substrate, regardless of the shape of the pattern surface.

4. Test Example in which Rf-POSS1 is used for one line pattern surface

Example 18 and Comparative Example 6

[0100] On the same silicon wafer as in Preparation Example 1, a fluorine organic silane Rf-POSS1 shown in Table 4 was chemically adsorbed in a liquid phase by the same method as in Preparation Example 7. Then, the silicon wafer was irradiated with vacuum-ultraviolet light (172 nm) through a line-shaped photomask shown in Fig. 4 to form a silanol group in the irradiated area. Formation of the objective pattern was confirmed by FE-SEM. In Example 18, a 1 wt% poly(3-hexylthiophene) solution was dropped on the pattern surface by using an ink-jet apparatus produced by us. Ink-jet conditions are as follows. That is, a microdroplet having a diameter of several tens of μm was prepared one by one by an on-demand system and then dropped at intervals of 1 mm. In Comparative Example 6, the microdroplet was dropped by the ink-jet apparatus on Rh(C10)/SO₃H having a line width of 10 μm (Comparative Preparation Example 2 shown in Table 2) in the same manner as in Example 18. The shape of the poly(3-hexylthiophene) thin film formed on the pattern surface was observed by an optical microscope. As a result, it has been found that a thin film was formed in a line shape, which is faithful to the line shape having a line width of 10 μm , was formed in Example 18. The thin film was formed only at the SiOH region. In Comparative Example 6, a circular thin film having a diameter of about 100 μm was formed, regardless of lines of the pattern surface.

5. Test Example in which a fluorine compound having a specific structure is used for one pattern surface with various shapes

Example 19 and Comparative Example 7

[0101] On the same silicon wafer as in Preparation Example 1, a fluorine organic silane Rf-POSS1 shown in Table 4 was chemically adsorbed in a liquid phase in the same manner as in Preparation Example 7. Then, the silicon wafer was irradiated with vacuum-ultraviolet light (172 nm) through a line-shaped photomask shown in Fig. 4 to form a silanol group in the irradiated area. Formation of the objective pattern was confirmed by FE-SEM. In Example 19, using this pattern surface as a template, pentacene was vapor-deposited on the pattern surface (150 nm thick). In Comparative Example 7, pentacene was vapor-deposited on an untreated silicon wafer (160 nm thick). In both cases, pentacene was vapor-deposited at a substrate temperature of 200°C.

[0102] After placing a metal deposition mask on the substrate, source and drain electrodes of gold were formed on the substrate by a vacuum deposition method. The source and drain electrodes were disposed vertically to lines of the substrate. The thickness of the gold vapor-deposited film was set to 100 nm. A distance (channel length) L between the source and drain electrodes was set to 10 μm and the length (channel width) W of the source and drain electrodes was set to 10 mm. Then, an Al thin film for leading-out a gate electrode was vapor-deposited on the back surface of the silicon wafer to give an organic TFT comprising a pentacene vapor-deposited film.

[0103] Mobility was 0.13 cm²/Vs in case of using the pattern surface as the substrate, while mobility was 0.005 cm²/Vs in case of using the untreated silicon wafer as the substrate.

EFFECT OF THE INVENTION

[0104] According to the present invention, an anisotropic material having an alternating-line pattern structure can be produced by a simple application (or coating) process.

[0105] In the present invention, a structure of a functional compound of nanometer to micrometer order can be produced by a simple application process.

[0106] In the present invention, properties of a functional compound can be improved by using, as a template, the surface of a pattern, at least one region of which is surface-treated with a fluorine compound having a specific structure.

Claims

1. An anisotropic material comprising an alternating-line pattern and a layer of at least one functional compound

selected from the group consisting of a semiconductor compound, an electrically conductive compound, a photochromic compound and a thermochromic compound, formed on a surface of the alternating-line pattern, wherein one type of lines in the alternating-line pattern surface comprises a fluorine-containing compound or silicone.

2. The anisotropic material according to claim 1, wherein a difference between surface free energy of the type of lines comprising the fluorine compound or silicone and surface free energy of the other type of lines is at least 5 mJ/m².
3. The anisotropic material according to claim 1, wherein the alternating-line pattern has a line width of 0.5 to 100 μ m.
4. The anisotropic material according to claim 1, wherein the alternating-line pattern has unevenness of not more than 10 nm.
5. The anisotropic material according to claim 1, wherein the shape of droplets is distorted when 2 μ L of ethanol is gently dropped from above the alternating-line pattern, and the degree of distortion is at least 1.1 in terms of a ratio L/W of the length in a major axis (L) to the length in a minor axis (W) of droplets.
6. The anisotropic material according to claim 1, wherein the alternating-line pattern comprises an organic silane compound, an organic thiol compound, an organic disulfide compound and/or an organic phosphoric acid ester.
7. A method for producing an anisotropic material, which comprises applying a solution of at least one functional compound selected from the group consisting of a semiconductor compound, an electrically conductive compound, a photochromic compound and a thermochromic compound on the surface of an alternating-line pattern, one type of lines of which comprises a fluorine-containing compound or silicone.
8. The method according to claim 7, wherein a liquid which dissolves the functional compound is a solvent having a surface tension of not more than 30 mN/m.
9. A method for producing a functional material, comprising using, as a template, a pattern surface composed of plural regions each having different surface free energy,
characterized in that:
 - (1) at least one region of the pattern surface is treated with a fluorine compound, and
 - (2) the method comprises applying a functional compound solution on the pattern surface and removing a solvent.
10. The method according to claim 9, wherein the fluorine compound comprises a fluorine compound having the following structure:
 - (a) a fluorine compound which has a branched fluoroalkyl group having 5 or less carbon atoms,
 - (b) a fluorine compound having a perfluoropolyether group,
 - (c) a fluorine compound having a polymer structure obtained by polymerizing a monomer which has a fluoroalkyl group having 5 or less carbon atoms,
 - (d) a fluorine compound having a linking group which is any one of an urethane group, an ester group, an ether group and an amide group, existing between a fluoroalkyl group having 5 or less carbon atoms and a functional group,
 - (e) an incompletely-condensed silsesquioxane which has a fluoroalkyl group having 5 or less carbon atoms, and/or
 - (f) a completely-condensed silsesquioxane which has a silane group and a fluoroalkyl group having 5 or less carbon atoms.
11. A functional material produced by the method according to claim 9.
12. A method for producing a functional material, which comprises applying a functional compound to a pattern surface having at least one region surface-treated with a fluorine compound.
13. The method according to claim 12, wherein the fluorine compound comprises a fluorine compound having the following structure:

- (a) a fluorine compound which has a branched fluoroalkyl group having 5 or less carbon atoms,
(b) a fluorine compound having a perfluoropolyether group,
(c) a fluorine compound having a polymer structure obtained by polymerizing a monomer which has a fluoro-
alkyl group having 5 or less carbon atoms,
5 (d) a fluorine compound having a linking group which is any one of an urethane group, an ester group, an
ether group and an amide group, existing between a fluoroalkyl group having 5 or less carbon atoms and a
functional group,
(e) an incompletely-condensed silsesquioxane which has a fluoroalkyl group having 5 or less carbon atoms,
and
10 (f) a completely-condensed silsesquioxane which has a silane group and a fluoroalkyl group having 5 or less
carbon atoms.

14. A functional material produced by the method according to claim 12.

15

20

25

30

35

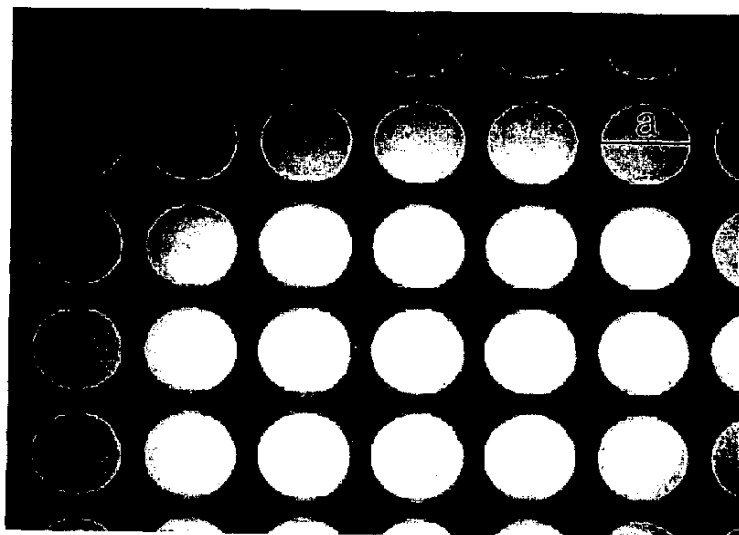
40

45

50

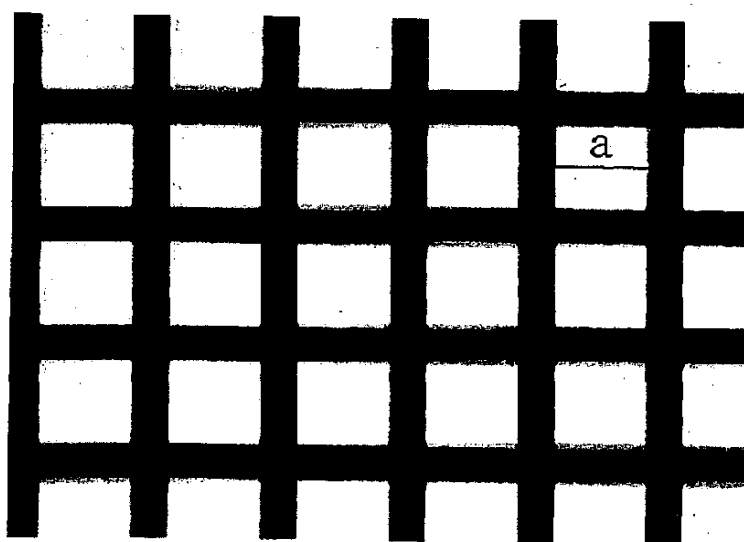
55

Fig.1



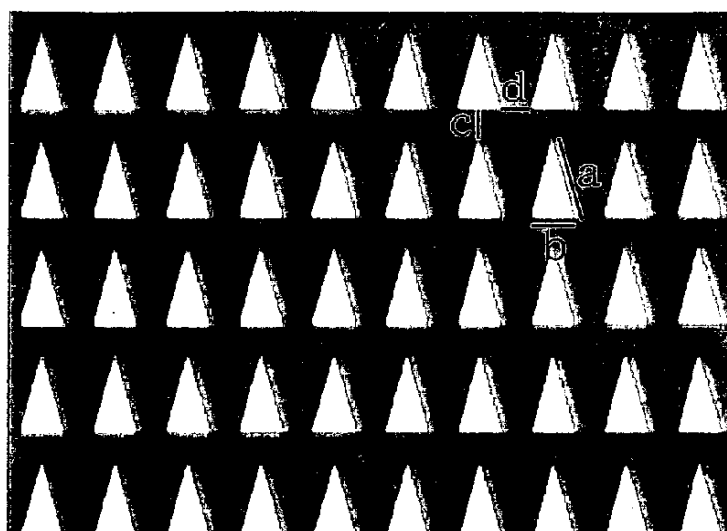
a : 170nm
b : 30nm

Fig.2



a : 16.7 μ m
b : 6.7 μ m

Fig.3



a : $15\mu\text{m}$
 b : $8.3\mu\text{m}$
 c : $6.7\mu\text{m}$
 d : $5\mu\text{m}$

Fig.4



a : $10\mu\text{m}$
 b : $10\mu\text{m}$

INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP03/11876

A. CLASSIFICATION OF SUBJECT MATTER Int.Cl. ⁷ H01L51/00		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) Int.Cl. ⁷ H01L51/00		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Jitsuyo Shinan Koho 1922-1996 Jitsuyo Shinan Toroku Koho 1996-2003 Kokai Jitsuyo Shinan Koho 1971-2003 Toroku Jitsuyo Shinan Koho 1994-2003		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	JP 2002-23356 A (Seiko Epson Corp.), 23 January, 2002 (23.01.02), Full text; Figs. 1 to 5 (Family: none)	1-14
X	JP 2001-284289 A (Seiko Epson Corp.), 12 October, 2001 (12.10.01), Full text; Figs. 1 to 7 (Family: none)	1-14
X	EP 1041652 A2 (LUCENT TECHNOLOGIES INC.), 04 October, 2000 (04.10.00), Full text; Fig. 1 & JP 2000-307172 A Full text; Fig. 1 & KR 1006904 A & US 2001/0029103 A1	1-14
<input checked="" type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.		
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed "W" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family		
Date of the actual completion of the international search 17 December, 2003 (17.12.03)		Date of mailing of the international search report 13 January, 2004 (13.01.04)
Name and mailing address of the ISA/ Japanese Patent Office		Authorized officer
Facsimile No.		Telephone No.

Form PCT/ISA/210 (second sheet) (July 1998)

INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP03/11876

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	JP 2002-261048 A (Seiko Epson Corp.), 13 September, 2002 (13.09.02), Full text; Figs. 1 to 7 & US 2003/0024103 A1	1-14
X	JP 2001-284288 A (Seiko Epson Corp.), 12 October, 2001 (12.10.01), Full text; Figs. 1 to 2 (Family: none)	1-14
X	JP 2001-284274 A (Seiko Epson Corp.), 12 October, 2001 (12.10.01), Full text; Figs. 1 to 6 (Family: none)	1-14
P,X	JP 2002-311592 A (Seiko Epson Corp.), 23 October, 2002 (23.10.02), Full text; Figs. 1 to 19 & US 2002/0182890 A1	1-14
P,X	JP 2003-162058 A (Seiko Epson Corp.), 06 June, 2003 (06.06.03), Full text; Figs. 1 to 14 (Family: none)	1-14

Form PCT/ISA/210 (continuation of second sheet) (July 1998)